Performance of RiceHusk A Geopolymer Concrete

1Suprava Samal, 1Bhabani Sankar Sahani, 1 Sanghamitra Mohapatra, 1Suman Kumar Behera, 2Swaraj Saurav

1Assistant Professor, Gandhi institute of technology and management, Bhubaneswar

2 Student of Department of Electrical & Electronics Engineering GITAM, BHUBANESWAR ODISHA

Abstract: Environmental issues resulted from cement production have become a major concern today. To develop a sustainable future it is encourage tol imit the use of this construction material that can affect the environment. Geopolymer is the best solute on reduce the use of cement in concrete. Geopolymer is the hardened cementiouspaste made from flyash, alkaline solution nand geological source material .Fly ash is finely divided powder produced by coal and fired power station .RHA is the bye-product of paddy industry. RHA is rich in silicaabout90%,5%carbonand 2%K2O.ThespecificsurfaceareaofRHAisabout40-100 m2/g. Asper study it is found that incorporation of RHA upto 30% replacement level reduces the chloride penetration, decreases permeability and strength and corrosion resistance properties.

Keywords: Physio-Chemical Analysis, Ground water samples, Landuseand Salem

INTRODUCTION

Geopolymer are inorganic materials that form long range, covalently bonded, nonnetworks. crystalline Commercially produced geopolymers may be used for fireand heat resistant coatings and adhesives. According to T.F. Yen. geopolymers can be classified in to two major groups: pure inorganic geopolmers and organic containing geopolymers of naturally occurring macromolecules. The main constituents of geopolymers, namely source material and the alkaline liquids. The source materials for geopolymers based onalumino-silicate should be richinsilicon(Si) and Aluminium(Al).Usuallyit would be byproducts suchas flyash, silica fume, rice husk ash, GGBS, red mud etc., The choice of source material for making geopolymer depends on factors such as availability, cost, type, application and specific demand of the end users.

The most common alkaline liquids used ingeo-

polymerisation is a combination of sodium hydroxide (NaOH) or potassium hydroxide (KOH) and sodium silicate or potassium silicate. Geopolymer concrete can be produced from ground granulated blast furnace slag (GGBS) and alkaline liquids. Water is added to the concrete only for workability and easiness in placing the concrete. On the other hand, in Portland cement, water is necessary for the hydration of cement. The fundamental unit within age of polymer structure is tetrahedral complex on sisting of Si or Al coordinated through covalent bonds of four oxygens. The geopolymeric framework results from the cross-linking between these tetrahedral, which leads to a 3-dimensional aluminosilicate network where the negative charge associated with tetrahedral aluminium is balanced by a small cationic species, most commonly an alkali metal cation.

II. COLLECTION OF LITERATURE

Greenhouse effect is a natural phenomenon. It refers to the atmosphere capability to prevent part of sun radiance to go back into space.The mechanism insurest hate art have rage temperature is kept between-18°Cand15°C.Gases which allow this phenomenon

arenotnaturallyhighlyconcentratedintheatm ospherebuthumanactivitieshavebeenincreas ingthoseconcentrations.Therefore

atmosphere characteristics and greenhouse effect has been altered. This alteration often considered as the main cause of global warming. Over the last century, carbon dioxide concentration in the atmosphere has increased of 30% and it is estimated that in 2013, CO2, with approximately 467 million tonnes (Mt), represented 83% of UK's greenhouse emissions. In general, this report will not deal with other gas but whenCO2e will be considered, they are considered to be included. It is this massive participation

That makes of CO2 the priority of any policya imingare duction of green house emissions. However, because their global warming potential is greater than the one of CO2, these other gas must be kept in mind and further studies might be carried out.

Increasingeconomicgrowthdemandsmoreef fectiveutilizationofbothrenewableandnonrenewableresources.Ifnotcarefully targeted, this will lead to production of an increased amount of waste. Additionally, waste handling companies treat numerous waste flows generated in different industrial areas which have no tyet found major industrial applications. Legislati on and taxation are becoming more stringent, and consequently, waste disposal costs are increasing. This has resulted in increasing interest in the search for novel utilization possibilities for the waste flows generated.

III. NUMERICAL ANALYSIS

The various mix combination sthat are to be casted areF90R10,F80R20,F70R30using three different samples of rice husk ash. In

this research work the compressive strength of Geopolymer concrete is examined for the mixes of 10 Molarity of sodium hydroxide. Themolecularweightofsodiumhydroxideis4 0.Toprepare10Molarityofsolution400gofso diumhydroxideflakesareweighed andtheycanbedissolvedindistilledwatertofo rm1litresolution.Volumetricflaskof1literca pacityistaken,sodium hydroxide flakes are added slowly to distilled water to prepare 1liter solution.

IV. THEORETICAL ANALYSIS

As there are no code provisions for the mix design of geopolymer concrete, the density of geo-polymer concrete is assumed as 2400 Kg/m3. The rest of the calculations are done by considering the density of concrete. The total volume occupied by fine and coarseaggregateisadoptedas77%. The alkaline liquid to flyash and GGB Sratioiskeptas0.4. The ratio of sodium hydroxideto sodium silicate is kept as 2.5. The conventional method used in the making of normal concrete is adopted to prepare geopolymer concrete.

V. EXPERIMENTAL TEST

Firstly, the fine aggregate, coarse aggregate, fly ash and GGBS are mixed in dry condition for 3-4minutes and then the alkaline solution which is combination of Sodium hydroxide solution and Sodium silicate solution with super-plasticizer is added to the dry mix. Water is taken as 10 % of the Cementous material (fly ash and GGBS). The super plasticizer is taken as 3% of the Cementous material. The mixing is done for about 6-8 mins for proper bonding of all the materials. After the mixing is done, cubes are casted by giving proper compaction in three layers.

Thespecimenistobeleftundisturbedforabout 48hoursbecauseof longer setting time.

The following are the materials required for 1m3 of concrete:



Fig. 1:Mixing of binders and aggregates

VII. RESULTS AND DISCUSSION

CONCLUSION

TheuseofRHAaspartialreplacementofceme ntinmortarandconcretehasbeenextensivelyi nvestigatedinthispaper.The following are the conclusions that are drawn from this study: Concrete requires approximate increase in water cement ratio due to increase in percentage of RHA. Because RHA is highly porous material. The workability of RHA concrete has been found to decrease within crease in RHA replacement. It was found thatricehusk when burnt produced amount of silica(morethan80%).Forth is reason it provides excellent thermal insulation. Rice contains moresilica. huskash and hencewepreferricehuskashuseinconcretetha nsilicafumetoincreasethestrength.Though Ricehusk as his harmful for humanbeing,

but the cost of ricehusk as his zero and thus wepreferRHAuseinconcrete as compared to silica fumes.

REFERENCES

[1] Malhotra, V. M., "Introduction: Sustainable Development and Concrete Technology," Concrete International, V. 24, No. 7, July 2002, p. 22. [2] Mehta, P. K., and Burrows, R. W., "Building Durable Structures in the 21st Century," Concrete International, V. 23, No. 3, Mar. 2001, pp. 57-63.

[3] Mehta, P. K., "Reducing the Environmental Impact of Concrete,"

Concrete International, V. 23, No. 10, Oct. 2001, pp. 61-66.

[4] Mehta, P. K., "Greening of the Concrete Industry for Sustainable Development," Concrete International, V. 24, No. 7, July 2002, pp. 23-28.

[5] Malhotra, V. M., "High-Performance High-Volume Fly Ash Concrete,"

Concrete International, V. 24, No. 7, July 2002, pp. 1-5.

[6] Davidovits, J., "Chemistry of Geopolymeric Systems, Terminology,"

Geopolymer '99 International Conference, France, 1999, pp. 9-40.

[7] Malhotra, V. M., "Making Concrete 'Greener' with Fly Ash," Concrete International, V. 21, No. 5, May 1999, pp. 61-66.

[8] ASTM C 618, "Standard Specification for Coal Fly Ash and Raw or Calcined Natural Pozzolan for Use as a Mineral Admixture in Concrete," ASTM International, West Conshohocken, Pa., 2001, 4 pp. [9] Palomo, A.; Grutzeck, M. W.; and Blanco, M. T., "Alkali-Activated Fly Ashes, A Cement for the Future," Cement and Concrete Research, V. 29, No. 8, 1999, pp. 1323-1329.

[10] van Jaarsveld, J. G. S.; van Deventer, J. S. J.; and Lukey, G. C., "The Effect of Composition and Temperature on the Properties of Fly Ash and Kaolinite-Based Geopolymers," Chemical Engineering Journal, V. 89, No. 1-3, 2002, pp. 63-73.

[11] Barbosa, V. F. F.; MacKenzie, K. J.
D.; and Thaumaturgo, C., "Synthesis and Characterization of Materials Based on Inorganic Polymers of Alumina and Silica: Sodium Polysialate Polymers," International Journal of Inorganic Materials, V. 2, No. 4, 2000, pp. 309-317.

[12]

Davidovits.J(1994)"Globalwarming impactonthecementandaggregates industrie s",WorldResourceReview,6,263-273.

[13]

Davidovits, J. (1999). "ChemistryofG eopolymericSystems, Terminology". Geopo lymer99thInternationalConference, France.

[14] DavidovitsJ.,(1993),Carbon-DioxideGreenhouse-

Warming:WhatFutureforPortlandCement,P roceedings,EmergingTechnologiesSymposi umonCement and Concretes in the Global Environment, 21p, Portland Cement Association, Chicago, Illinois, March 1993.

[15] AS 1012.9-1999, "Determination of the Compressive Strength of Concrete Specimens," Australian Standard, 1999, 10 pp.

[16] Warner, R. F.; Rangan, B. V.; Hall,A. S.; and Faulkes, K. A., ConcreteStructures, Melbourne: Addison WesleyLongman Australia Ltd., 1998, 974 pp.

[17] Wallah, S. E.; Hardjito, D.; Sumajouw, D. M. J.; and Rangan, B. V., "Sulfate Resistance of Fly Ash-Based Geopolymer Concrete," Concrete in The Third Millenium, 21st Biennial Conference of The Concrete Institute of Australia, Brisbane, Queensland, Australia, 2003, pp. 205-212.